

the buffer containing 0.2 M NaCl to give the xylan-containing fraction. Since the preliminary experiment revealed that the xylan became partly water-insoluble after freeze-drying, the isolation procedures were performed without lyophilization. The 0.2 M NaCl fraction gave, on complete acid hydrolysis, L-arabinose, D-xylose and D-galactose in the molar ratio 38.3:33.4:28.3 with trace amounts of D-mannose and D-glucose as neutral sugars. This fraction, after dialysis against H₂O, was applied to a column (2.3 × 12 cm) of cellulose powder (Whatman, CF 11), which was first washed with water to give crude AGP and then eluted with 1 M NaOH to give acidic xylan. The alkaline fraction was dialysed against H₂O and then centrifuged. The supernatant gave L-arabinose, D-xylose, D-mannose, D-glucose and D-galactose in the molar ratio 4.0:89.5:1.3:3.1:2.1. The yield of the acidic xylan was ca 50 mg from 1 l of ECP soln.

REFERENCES

1. Aspinall, G. O. (1980) in *The Biochemistry of Plants* (Preiss, J., ed.), Vol. III, p. 473. Academic Press, New York.
2. Eda, S., Ohnishi, A. and Katō, K. (1976) *Agric. Biol. Chem.* **40**, 359.
3. Eda, S., Watanabe, F. and Katō, K. (1977) *Agric. Biol. Chem.* **41**, 429.
4. Akiyama, Y. and Katō, K. (1982) *Phytochemistry* **21**, 1325.
5. Akiyama, Y. and Katō, K. (1981) *Phytochemistry* **20**, 2507.
6. Yarov, J., Rapport, M. M. and Graf, L. (1962) *Biochem. J.* **85**, 383.
7. Kováč, P., Hirsch, J., Shashkov, A. S., Usov, A. I. and Yarotsky, S. V. (1980) *Carbohydr. Res.* **85**, 177.
8. Shashkov, A. S., Sviridov, A. F., Chizhov, O. S. and Kováč, P. (1978) *Carbohydr. Res.* **62**, 11.
9. Aspinall, G. O., Molloy, J. A. and Craig, J. W. T. (1969) *Can. J. Biochem.* **47**, 1063.
10. Dubois, M., Gilles, K. A., Hamilton, J. K., Rebers, P. A. and Smith, F. (1956) *Analyt. Chem.* **28**, 350.
11. Blumenkrantz, N. and Asboe-Hansen, G. (1973) *Analyt. Biochem.* **54**, 484.
12. Albersheim, P., Nevins, D. J., English, P. D. and Karr, A. (1967) *Carbohydr. Res.* **5**, 340.
13. Hakomori, S. (1964) *J. Biochem.* **55**, 205.
14. Akiyama, Y., Mori, M. and Katō, K. (1980) *Agric. Biol. Chem.* **44**, 2487.
15. Lowry, O. H., Rosebrough, N. J., Farr, A. L. and Randall, R. J. (1951) *J. Biol. Chem.* **193**, 265.
16. Akiyama, Y., Eda, S., Mori, M. and Katō, K. (1983) *Phytochemistry* **22**, 1177.

Phytochemistry, Vol. 23, No. 9, pp. 2062–2063, 1984
Printed in Great Britain

0031-9422/84 \$3.00 + 0.00
© 1984 Pergamon Press Ltd

VOLATILE COMPONENTS IN CELL SUSPENSION CULTURES OF *CRYPTOMERIA JAPONICA*

NARIYUKI ISHIKURA*, KENSUKE NABETA† and HIROSHI SUGISAWA†

Department of Biology, Faculty of Science, Kumamoto University, Kurokami, Kumamoto 860, Japan; †Department of Food Science, Faculty of Agriculture, Kagawa University, Kagawa 761-07, Japan

(Received 9 December 1983)

Key Word Index—*Cryptomeria japonica*; Taxodiaceae; cell suspension culture; volatiles; diterpenes; aldehydes; fatty acids and their esters.

Abstract—The cell suspension culture of *Cryptomeria japonica* contains volatile oils, the yield of which was 0.005–0.01% of the fresh cells. In the volatiles, five aldehydes, ten fatty acids and their esters, and two diterpenes of abietatriene and ferruginol have been found. Of these, palmitic acid is present as the most predominant component, amounting to ca 40% of the volatiles.

INTRODUCTION

Cryptomeria is a rich source of volatile oils. *C. japonica* (Japanese cedar) is known to yield ca 0.7% of volatiles from the fresh leaves, in which diterpenes of kaurene, isokaurene, 9-ketoferruginol (sugiol) and the others have been found [1]. Recently, many callus tissues and suspension cells have also been found to produce mono- and sesquiterpenes [2, 3]. As one of us recently succeeded in

establishing the cell culture of Japanese cedar [4], we thus examined the volatile oil in the suspension cultured cells.

RESULTS AND DISCUSSION

The cell suspension culture of Japanese cedar was maintained in a modified liquid medium based on that of Schenk and Hildebrandt [6]. From this, the 14th generation of suspension cells was harvested at a 14-day culture (the stationary growth phase) and subjected to extraction of volatiles. The yield of volatile oils obtained by a simultaneous distillation extraction [7] was 0.005–0.01% of the fr. wt. The cells at this growth stage contained a

*To whom correspondence should be addressed.

Table 1. Volatile components in cell suspension cultures of *Cryptomeria japonica*

Compounds	GC R _i (min)	Approx. % of volatiles*
Aldehydes		
Nonanal	1.6	0.5
(2E,4Z)-2,4-Heptadienal	2.2	1.2
(2E,4E)-2,4-Heptadienal	2.3	1.0
2,4-Nonadienal	4.3	0.6
(2E,4Z)-2,4-Decadienal	5.2	1.2
(2E,4E)-2,4-Decadienal	5.5	3.2
Acids		
Myristic acid	15.5	0.6
Palmitic acid	17.6	38.5
Linoleic acid	22.0	2.1
Linolenic acid	23.3	0.5
Esters		
Methyl palmitate	11.1	3.5
Ethyl palmitate	11.4	4.4
Methyl linoleate	13.6	0.6
Ethyl linoleate	14.2	4.5
Methyl linolenate	14.3	0.8
Ethyl linolenate	14.6	5.1
Diterpenes		
Abietatriene	13.1	1.6
Ferruginol	26.2	1.6

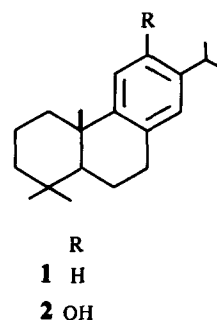
*Calculated from relative peak height by GC.

maximal level of volatiles, and also, as reported in a previous study [4], they contained a small amount of chloroplasts and a large amount of catechin and procyanidins.

By GC/MS analysis of the volatile oils from the suspension cells (Table 1), five aldehydes, 10 fatty acids and their esters, and two diterpenes were identified as the major components. In particular, the cells yielded large amounts of palmitic acid, which comprised ca 40% of the volatiles. We also found that the cells produced seven diterpenes, of which abietatriene (1) and ferruginol (2) were identified. By a preliminary examination, both compounds also have been confirmed to be present in the green leaves of Japanese cedar. 9-Ketoferruginol, which have been reported as a characteristic component in the leaves, could not be detected in the suspension cells.

EXPERIMENTAL

Plant material. *C. japonica* cv 'kumotôshi' used for the present study is a tall tree growing in Japan and produces valuable timber. As described previously [4], the callus was induced from the tip section of a young shoot with Murashige and Skoog medium [5] containing 2,4-D (1 mg/l), kinetin (0.1 mg/l), sucrose (30 g/l) and agar (0.9% w/v). The callus tissues were then subcultured every 3 weeks for ca 5.5 months at 25° in the dark. From calluses at the 8th generation, the cell suspension culture was derived and maintained in a modified Schenk and Hildebrandt medium [6] containing 2,4-D (0.5 mg/l), kinetin (0.01 mg/l) and sucrose (30 g/l). A portion (1.2–1.4 g) of the suspension cells in a Erlenmeyer flask (200 ml) was regularly transferred to fresh medium (50 ml each) every 2 weeks. Flasks were agitated on a



reciprocal shaker at 62 strokes per min with horizontal excursion of 5.5 cm at 25° under white light (350 lx). As estimated from the increase in fr. wt, the doubling time of the cells in the logarithmic growth phase (9 days after the 3–4 days lag growth phase) was ca 48 hr. The suspension cells of the 14th generation (the 22nd generation after callus formation) were harvested at a 14-day culture (the stationary growth phase) and subjected to extraction of volatiles.

Extraction and GC and GC/MS. Volatile oils from suspension cells were separated by a simultaneous distillation extraction [7]. The cells (41.5 g) harvested were homogenized in H₂O (300 ml) at ca 0° with a blender for 5 min. The homogenate was transferred into the vessel for a simultaneous distillation extraction. The soln obtained was extracted with 30 ml of Et₂O for 90 min. Five ml of tridecane in 5 µl of Et₂O was then added to the Et₂O extract. After concn of the Et₂O extract under N₂, and under red. pres. to ca 30 µl, the volatile components were analysed by GC and GC/MS as follows. For GC analysis, the column used was a 30 m × 0.3 mm WCOT glass capillary column coated with Carbowax 20 M. Column temp. 70–150° at 5°/min, injector temp. 170°, detector temp. 170°, N₂ carrier at 0.78 ml/min, split ratio 1:85. For GC/MS analysis, the column used was a WCOT glass capillary column coated with FFAP (50 m × 0.25 mm i.d.). Column temp. 40–210° at 3°/min, He carrier at 1.0 ml/min, split ratio 1:20. Ionizing voltage for both EIMS and CIMS was 70 eV; reagent gas for CIMS was CH₄.

Volatile components. Volatile components listed in Table 1 were identified by direct comparison with authentic specimens using GC/MS. MS of diterpenes were as follows. Abietatriene EIMS *m/z* (rel. int.): 270 [M]⁺ 255 (100), 213 (7), 199 (11), 185 (35), 173 (60), 159 (63), 143 (17), 141 (12), 129 (20), 117 (15), 91 (10), 83 (9), 69 (45). CIMS *m/z*. 271 [M + H]⁺ (70), 270 [M]⁺ (80), 269 [M – H]⁺ (70). Ferruginol EIMS *m/z* (rel. int.): 286 [M]⁺ (95), 272 (20), 271 (100), 229 (16), 215 (16), 201 (39), 189 (66), 175 (64), 69 (51), 55 (16); CIMS *m/z*: 287 [M + H]⁺ (70), 286 (100).

REFERENCES

- Hegnauer, R. (1962) *Chemotaxonomie der Pflanzen* Band 1. Birkhäuser, Basel.
- Nabeta, K. and Sugisawa, H. (1983) in *Instrumental Analysis of Foods*, Vol. 1 (Charalambous, G. and Inglett, G., eds) p. 65. Academic Press, New York.
- Takeda, R. and Katoh, K. (1981) *Planta* **151**, 525.
- Ishikura, N. and Teramoto, S. (1983) *Agric. Biol. Chem.* **47**, 421.
- Murashige, T. and Skoog, F. (1962) *Physiol. Plant.* **15**, 473.
- Mitchell, J. P. and Gildow, F. E. (1975) *Physiol. Plant.* **34**, 250.
- Shultz, T. H., Flath, R. A. and Mon, T. R. (1977) *J. Agric. Food Chem.* **25**, 446.